





**Climate, Climate Change
Nuclear Power and the
Alternatives**

PHYC 40050

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School of Mathematical Sciences
University College Dublin**

Introduction to Meteorology & Climate





Lecture 1

**The Composition and Structure
of the Atmosphere**



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OUR HOME

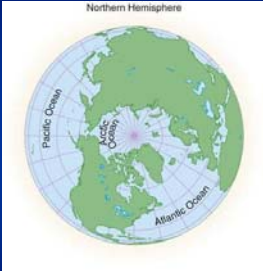



Earth (14C) 

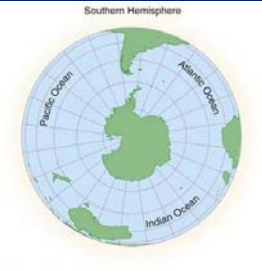
Moon (-18C) 

COMPARISON OF HEMISPHERES


Northern Hemisphere




Southern Hemisphere



70% of the globe covered by water



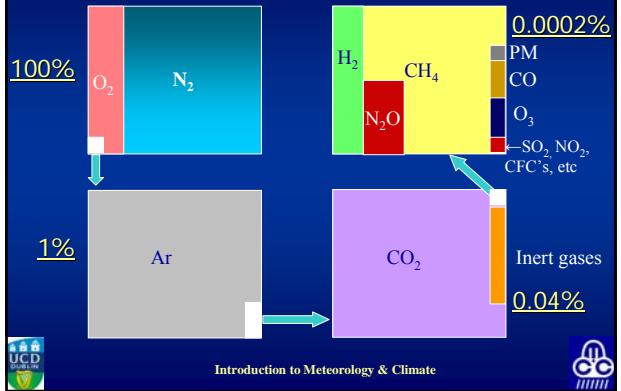
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Atmospheric Composition

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COMPOSITION OF THE EARTH'S ATMOSPHERE



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TABLE 1.1
Composition of the Atmosphere

Gas	Percentage by Volume
Nitrogen	~78% 78.08
Oxygen	~21% 20.95
Argon	~1% 0.93
Trace Gases	
Carbon dioxide	.04% 0.038
Methane	0.00017
Ozone	0.000004
Chlorofluorocarbons	0.00000002
Water vapor	Highly variable (0-4%)

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ATMOSPHERIC COMPOSITION

- Molecular **oxygen** and **nitrogen** are major components – 99%
- Of the remaining 1% , 96% is the inert gas argon
- Of the remaining 4%, 93% is carbon dioxide
- All remaining gases – about 2 parts in 100,000 are known as **trace species**
- These gases control the chemistry of the troposphere

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THE EARLY ATMOSPHERE

4.6 billion years ago:

- Earth's gravity too weak to hold hydrogen and helium (unlike Sun, Jupiter, Saturn, Uranus)
- Earth's present atmosphere from volcanoes (**outgassing**)
 - Water vapor condensed to form oceans
 - CO₂ went into oceans and rocks
 - N₂
- Oxygen forms by break-up of water by sunlight, later from plants ... **photosynthesis**.



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ATMOSPHERIC EVOLUTION

- Earth's **early atmosphere** consisted of Hydrogen (H), Helium (He), Methane (CH₄) and Ammonia (NH₃)
- As the earth cooled **volcanic eruptions** occurred emitting water vapour (H₂O), carbon dioxide (CO₂) and nitrogen (N₂).
- The molecular oxygen (O₂) in the current atmosphere came about as **single celled algae** developed in the oceans about 3 billion years ago.

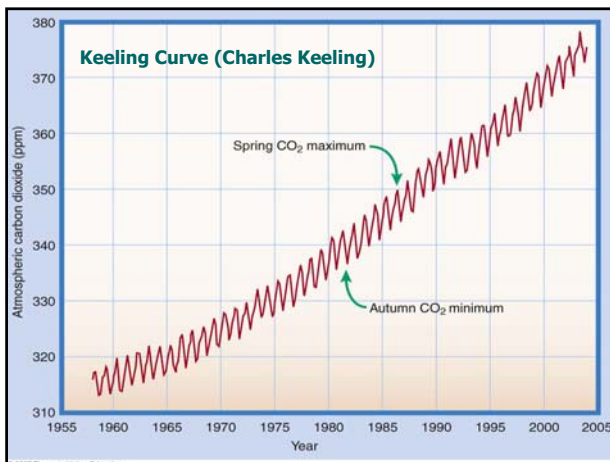
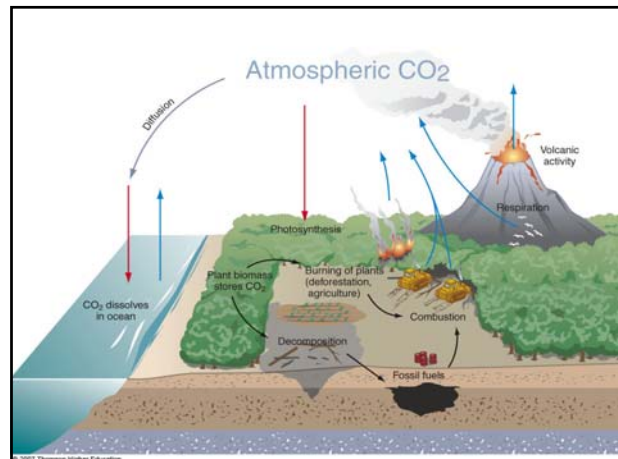
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ATMOSPHERIC EVOLUTION

- Oxygen is produced as a by-product of **photosynthesis**, the making of sugars from water vapor and carbon dioxide.
- This oxygen produces **ozone (O₃)** in the upper atmosphere which filtered out harmful ultraviolet radiation from the sun.
- This allowed **plants and animals** to develop on land.



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CARBON DIOXIDE CYCLE

Sources

- Plant/animal respiration
- Plant decay
- Volcanoes
- **Burning of fossil fuels**
- **Deforestation**

Sinks

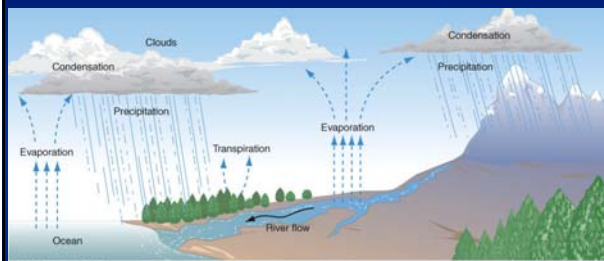
- Plant photosynthesis
- Oceans
- Carbonates



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HYDROLOGICAL CYCLE



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HYDROLOGICAL CYCLE

- Water is everywhere on earth
- It is in the oceans, glaciers, rivers, lakes, the atmosphere, soil, and in living tissue
- All these 'reservoirs' constitute the **hydrosphere**
- The continuous exchange of water between the 'reservoirs' is called the **hydrological cycle**
- The hydrological cycle is powered by the Sun
- It comprises
 - Evaporation and transpiration
 - Precipitation
 - Percolation into ground
 - Run-off to the sea



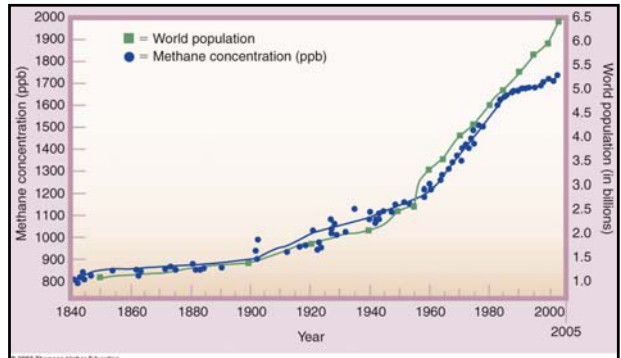
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Trace Constituents



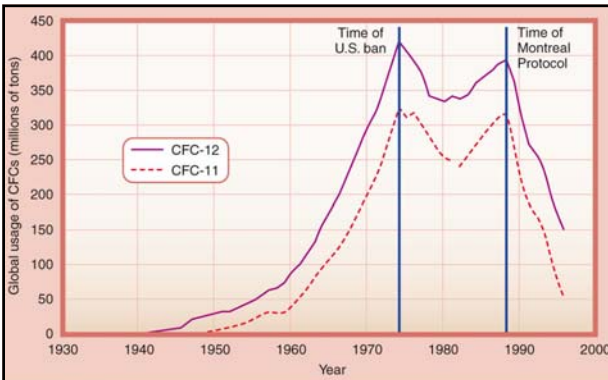
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Methane and world population



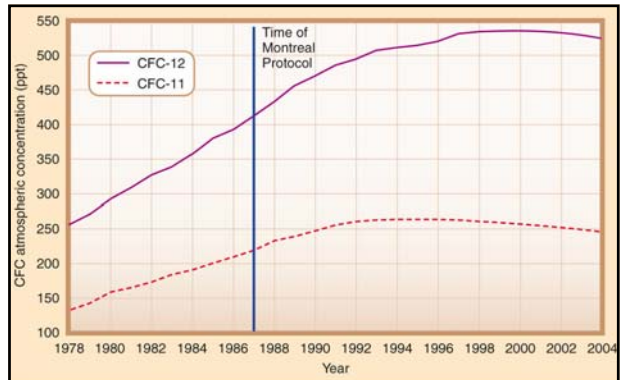
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CFCs: global production



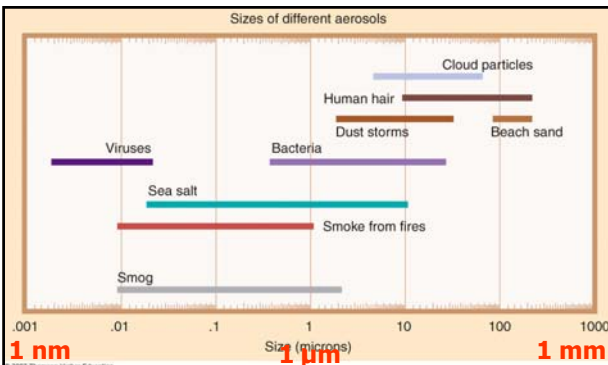
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CFCs: global concentration



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Aerosols: particle sizes



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AEROSOLS

- Particles suspended in the atmosphere
- Diameters of microns – one millionth of a meter.
- Modify the amount of solar energy reaching the surface.
- Act as **condensation nuclei** for cloud droplets.

PRIMARY SOURCES:

- Sea salt spray
- Wind erosion
- Volcanoes
- Fires
- Human activity



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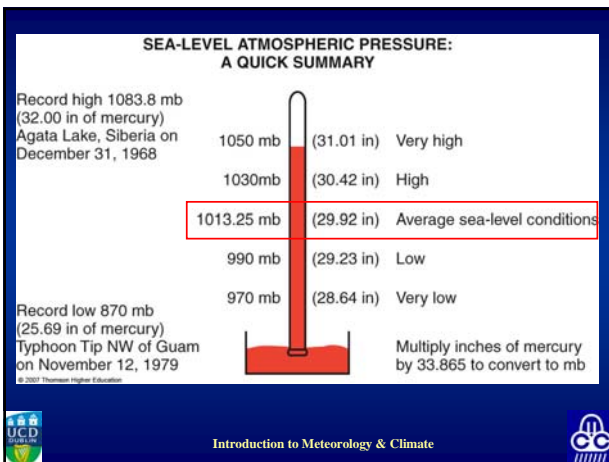


PRESSURE AND DENSITY

- **Pressure** is the force exerted on a given area.
- Air pressure results when air molecules move and collide with objects.
- Air pressure is exerted in all directions.
- **Density** is the concentration of molecules, or mass per unit volume.
- The pressure, density, and temperature of a gas are all related to each other.



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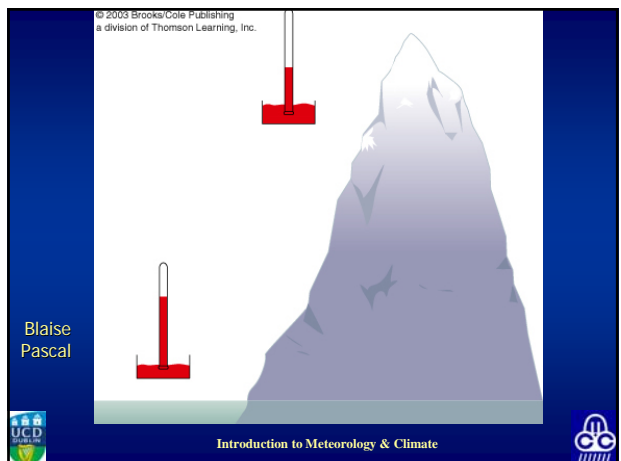
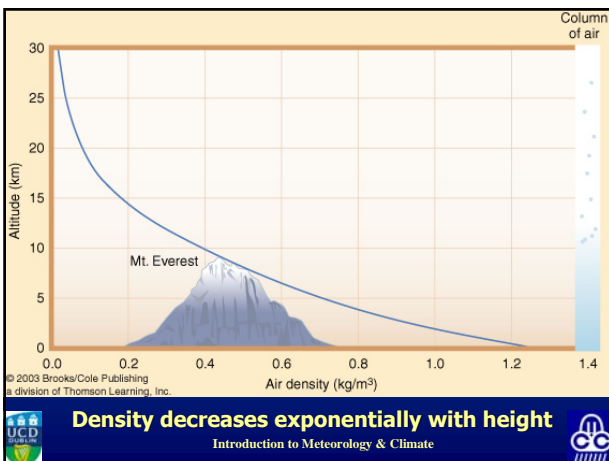


PRESSURE AND ALTITUDE

- Pressure is measured in terms of inches of mercury, or in **millibars** or **hectopascals**.
- Average sea-level pressure is 29.92 inches of mercury or 1013.25 millibars (hPa).
- Atmospheric pressure always decreases with increasing altitude.
- The air pressure measured on top of the Sugar Loaf is always less than the pressure in Kilmacanogue.
- To subtract the effect of station elevation, air pressure is corrected to report what it would be at sea level (**sea level pressure**)



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ATMOSPHERIC PRESSURE

- Pressure at a point is the **weight of air above that point**.
- A column of air of cross-section 1 square metre weighs about 10 tonnes !
- In still air, two factors determine the pressure – temperature and density
- **Ideal gas law:**
Pressure = Constant X Density X Temperature
- Pressure decreases with altitude.



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ATMOSPHERIC DENSITY

- The concentration of molecules is measured in terms of **density**, or mass per unit volume.
- Density at sea level for temperature of 15°C is about 1.2 kilograms per cubic metre.
- Density decreases with altitude.



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IDEAL GAS LAW

- The relationship between pressure, temperature, and volume is given by the ideal gas law:

$$p = R \rho T$$

where p = pressure
 R = the gas constant
 ρ = (Greek letter rho) density
 T = temperature



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IDEAL GAS LAW

- Knowing the Ideal Gas Law, you should be able to say what happens to one variable if a change in one of the others occurs (while the third remains constant).
- E.g., what happens to pressure if density increases (temperature constant)?



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STRATIFICATION OF THE ATMOSPHERE



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ATMOSPHERIC LAYERS

- The atmosphere can be divided up according to **pressure** (500 mb layer is about halfway up in the atmosphere).
- The atmosphere can also be divided up according to **temperature** (which does not follow a simple relationship with height).
- Averaging out temperature values in the atmosphere, we identify **four layers**.



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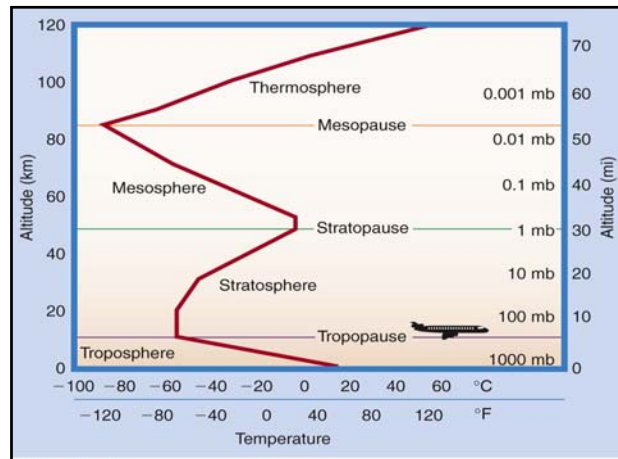


ATMOSPHERIC LAYERS

- **Troposphere** – temperature decreases with height
- **Stratosphere** – temperature increases with height
- **Mesosphere** – temperature decreases with height
- **Thermosphere** – temperature increases with height



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TROPOSPHERE

- From the surface up to about 12km (varies with latitude and season – higher in Summer, and in the tropics).
- Temperature decreases with height because the troposphere is heated by the surface and not directly by sunlight.
- **Almost all of what we call "weather" occurs in the troposphere.**
- Contains 80% of the atmosphere's mass



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STRATOSPHERE

- Between about 12km and 50km.
- Temperature increases with height because the **ozone layer** absorbs ultraviolet light and warms up as a result.
- Lack of mixing and turbulence.
- Very little exchange occurs between the stratosphere and troposphere (but it is important where it does).
- 99.9% of the atmospheric mass below the stratopause.



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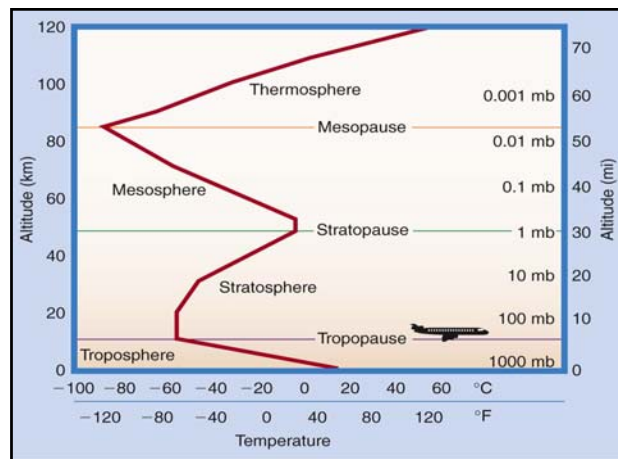


MESOSPHERE & THERMOSPHERE

- Mesosphere between 50km and 85km.
- Thermosphere goes up and up and up: there is no clear separation between the thermosphere and interplanetary space.
- The highest temperatures in the atmosphere are found in the thermosphere due to high energy radiation being absorbed by gases.
- Ionosphere (charged gas atoms) that reflects radio waves, and aurora are here.



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1. **Troposphere**- literally means region where air “turns over”
-temperature usually decreases (on average $-6.5^{\circ}\text{C}/\text{km}$) with altitude

Tropopause

2. **Stratosphere**- layer above the tropopause, little mixing occurs in the stratosphere, unlike the troposphere, where “turbulent mixing” is common

Stratopause

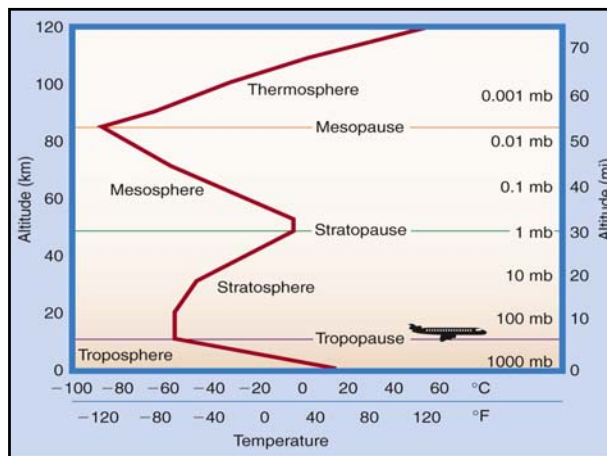
3. **Mesosphere**- defined as the region where temperature again decreases with height.

Mesopause

4. **Thermosphere**- region with very little of the atmosphere’s mass. High energy radiation received by the thermosphere: high temperatures. Very low density (not much “heat” felt).



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In meteorology we often refer to altitude as a certain pressure value rather than height.

The atmosphere moves mainly on constant pressure surfaces (isobaric surfaces)

- 850 mb \approx 1500 m (5000 ft)
- 700 mb \approx 3000 m (10,000 ft)
- 500 mb \approx 5500 m (18000)
- 300 mb \approx 9000 m (30,000)



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Introduction to Weather Maps



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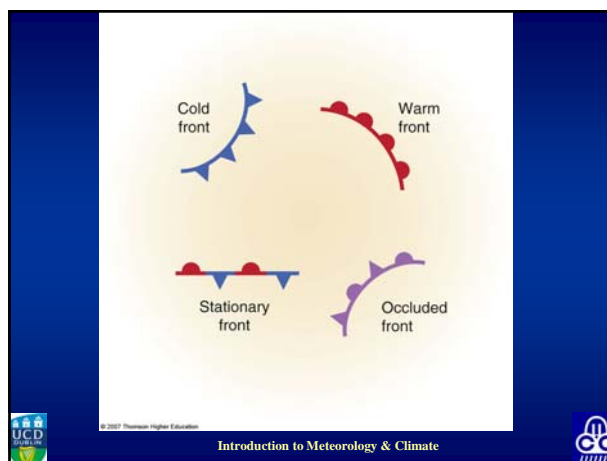


ATMOSPHERIC FRONTS

- **Front** – a boundary between two regions of air that have different meteorological properties, e.g. temperature or humidity.
- **Cold front** – a region where cold air is replacing warmer air.
- **Warm front** – a region where warm air is replacing colder air.
- **Stationary front** – a front that is not moving.
- **Occluded front** – a front where warm air is forced aloft.

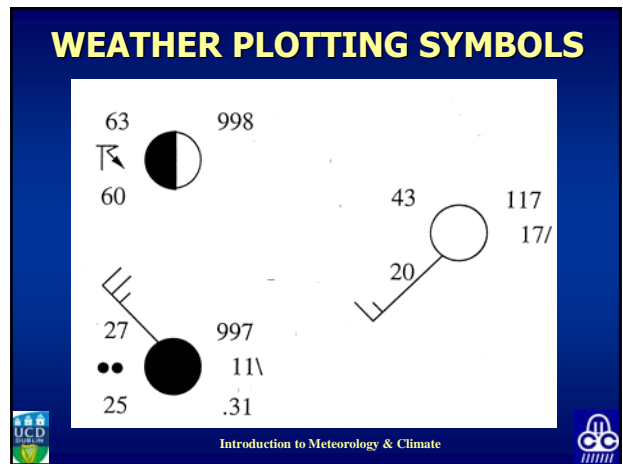
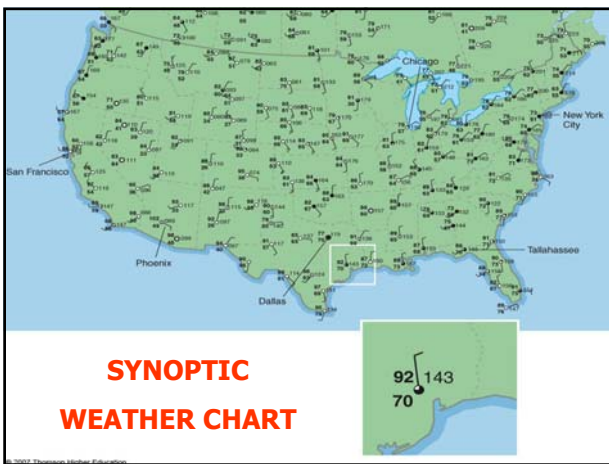
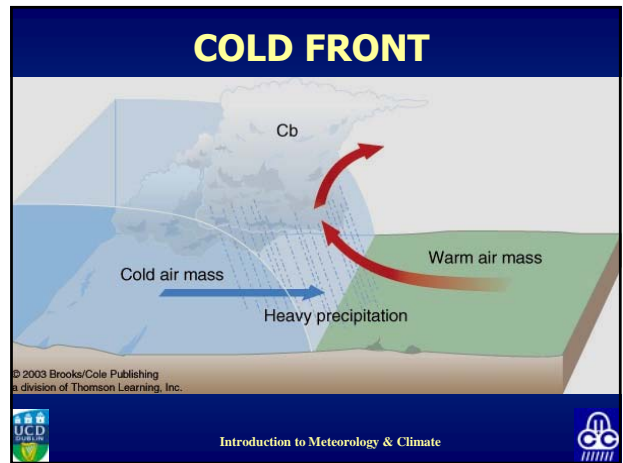
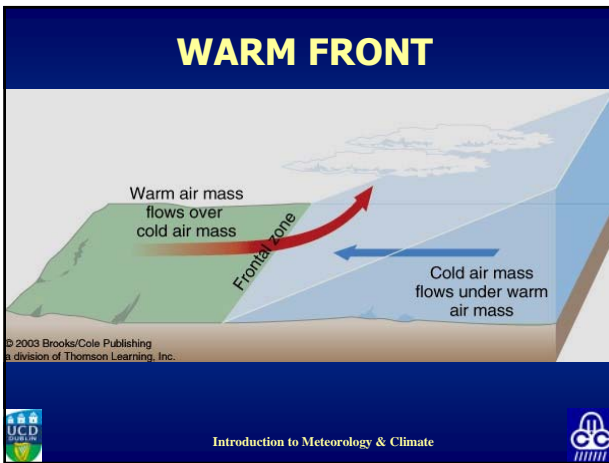


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Temperature		Wind direction	Wind speed	WIND SPEED	
Current weather	Dew point	Cloud cover	Pressure	miles per hour	kilometers per hour
76	55	138	-3	0	0
				1-2	1-3
				3-8	4-13
				9-14	14-19
				15-20	20-32
				21-25	33-40
				26-31	41-50
				32-37	51-60
				38-43	61-69
				44-49	70-79
				50-54	80-87
				55-60	88-96
				61-66	97-106
				67-71	107-114
				113-118	182-190
				119-123	191-198

RAIN	SNOW	DRIZZLE	FREEZING RAIN
Light	Light	Light	Light
Moderate	Moderate	Moderate	Moderate
Heavy	Heavy	Heavy	Heavy
Light shower	Light shower	Light	Light
Moderate shower	Moderate shower	Moderate	Moderate
Thunderstorm	OTHER		
Heavy T-storm	Haze	Ice crystals	
	Fog	Ice pellets (sleet)	

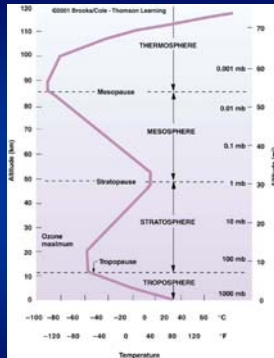
Cloud cover	Other
0% Cloud cover—clear skies	75% Cloud cover—broken clouds
10% Cloud cover—few clouds	90% Cloud cover—broken clouds
25% Cloud cover—few clouds	100% Cloud cover—overcast
40% Cloud cover—scattered clouds	⊗ Vision obscured
50% Cloud cover—scattered clouds	⊖ Missing data
60% Cloud cover—broken clouds	

METEOROLOGICAL TIME

- All weather reports are labeled using **Coordinated Universal Time (UTC)**, also called **Greenwich Mean Time (GMT)**, and also denoted **Zulu (Z)**.
- Zulu** is the time along the **0° longitude line**, which runs through **Greenwich**.
- Meteorology** uses the **24 hour clock** which omits the use of **a.m.** and **p.m.** (**0900 = 9 a.m., 2100 = 9 p.m.**)

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Revision: Layers of the Atmosphere



- Hot top: oxygen absorbs sunlight
- Warm middle: ozone absorbs ultraviolet (UV)
- Warm surface: land and ocean absorb sunlight



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End of Lecture 1



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